URS Greiner/Woodward Clyde A Division of URS Corporation

Hand Delivered

December 10, 1998

Mr. Gerald Delavan, P.G. Solid Waste Division Arkansas Department of Pollution Control and Ecology 8001 National Drive Little Rock, AR 72209

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Re: Work Plan for In-Situ Bioremediation At the El Dorado Chemical Company Plant El Dorado, Arkansas WC File 98B319 - 1

Dear Mr. Delavan:

Attached is one copy of the referenced report. We are looking forward to working with the Department on this important project.

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If you have questions or comments, I can be reached at (225) 756-1431.

Sincerely,

Cheryl Charen ODennis E. Reece Vice President

William Beal, P.G., P.E

cc: Mr. John Carver, LSB Industries Mr. Byron Smith, EDC



WORK PLAN

IN-SITU BIOREMEDIATION

RESPONSE TO PARAGRAPH 4(B) OF CONSENT ADMINISTRATIVE ORDER LIS 98-119

EL DORADO CHEMICAL COMPANY EL DORADO, ARKANSAS

Prepared for El Dorado Chemical Company El Dorado, Arkansas

December 1998

Woodward-Clyde

2822 O'Neal Lane Baton Rouge, LA 70816 (504) 751-1873 WC File 98B319

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APPENDICES

Appendix A Soil Boring Logs and Monitor Well Construction Diagrams for Monitor Wells to be Used in the In-Situ Bioremediation

El Dorado Chemical Corporation (EDCC) entered into Consent Administrative Order (CAO) LIS 98-119 with the Arkansas Department of Pollution Control and Ecology (ADPC&E) which became effective in October, 1998. The CAO requires EDCC to implement several actions. One of the provisions of the CAO (Paragraph 4, pages 15 and 16) requires an Interim Measure consisting of in-situ bioremediation of the existing groundwater monitoring wells which have exhibited nitrate concentrations in excess of 10 mg/L. The CAO requires EDCC to submit to the ADPC&E a work plan for in-situ bioremediation. This work plan is submitted in fulfillment of this requirement.

The CAO requires that this work plan describe the activities and present a schedule of significant dates for initiation of bioremediation, sampling the groundwater, submission of the sample analysis results to the ADPC&E, and preparation of annual reports evaluating the effectiveness of this Interim Measure. Section 2.0 of this Work Plan presents relevant background information for the facility. Section 3.0 describes the planned activities for the in-situ bioremediation and Section 4.0 presents the proposed implementation schedule. Section 5.0 describes the planned reporting of results. Section 6.0 lists references.

2.1 GENERAL FACILITY INFORMATION

The EDCC facility is located at 4500 North West Avenue in the city of El Dorado, Union County, Arkansas. The EDCC property consists of approximately 1,340 acres, of which about 150 acres are utilized for plant operations.

EDCC is a manufacturer of basic agricultural chemicals, including sulfuric acid, nitric acid, ammonium nitrate fertilizers and industrial grade ammonium nitrate. The facility is currently owned by El Dorado Chemical Company, a wholly owned subsidiary of LSB Industries of Oklahoma City, Oklahoma. EDCC purchased the plant in July 1983 from Monsanto Chemical Company, which had occupied the site since 1955. Previous site occupants included Lion Chemical Corporation (1943-1949) and Lion Oil Company (1949-1955). Based on information provided by EDCC, the plant property was undeveloped prior to 1943. Since 1943, site operations have generally been limited to production of ammonia-related products, sulfuric acid, and nitric acid.

EDCC is currently implementing procedures for pollutant source control, wastewater minimization and enhanced wastewater treatment/disposal measures. These measures are expected to reduce nitrate concentrations in the shallow groundwater.

2.2 REGIONAL GEOLOGIC AND HYDROGEOLOGIC SETTING

The EDCC facility lies within the Gulf Coastal Plain Province in southern Arkansas. This province is basically comprised of a relatively undissected low-lying plain underlain by complex interbedded sands, silts, and clays which thicken progressively toward the coast and toward the Mississippi River. Sediments within the sequence are, for the most part, unconsolidated or non-lithified. The sediments range in age from Quaternary (youngest) to Triassic.

Structurally, depositional dip was basinward in a general southern to southeasterly direction. Growth fault development at depth enhances the basinward dip of the sediment accumulations across the region. A graben structure (a down-thrown faulted block of sediments) is located approximately seven miles south of the facility. The fault planes which form the graben strike northwest-southeast.

SECTIONTWO

Table 2-1 shows the age relationships of the various formations found in the subsurface of the region. Also shown are the approximate thickness of each formation and a description of the hydrogeologic character of the sediments.

2.3 SITE SPECIFIC HYDROGEOLOGY

EDCC investigated the site hydrogeology with Phase I and Phase II groundwater investigations. The Phase I groundwater investigation drilled soil borings and installed temporary well points at thirty-five locations throughout the facility. The Phase I data were used to select locations for installation of monitor wells during the Phase II groundwater investigation. The Phase II groundwater investigation drilled 18 soil borings in which monitor wells were installed. The monitor wells were installed approximately 10 feet into the uppermost saturated unit with well depths ranging from 17 to 34.7 feet. Figure 2-1 shows the locations of the Phase II monitor wells. Reports dated January 1995 and January 1997 for the Phase I and II investigations, respectively, were prepared by Woodward-Clyde and submitted to the ADPC&E.

The soil borings indicate that the shallow soils at the site consist primarily of silty clays and clayey silts with occasional thin occurrences of silts or silty sands. Groundwater occurs within these deposits under conditions that range from semi-confined to unconfined.

Groundwater elevations were measured in the temporary well points during the Phase I groundwater investigation and in the monitor wells during the Phase II groundwater investigation and during a subsequent sampling event in October 1997. The results indicate the direction of shallow groundwater flow at the site is generally to the east-south-east, with the exception of areas locally influenced by ground surface topography and surface water. Figure 2-2 presents a groundwater elevation map for the most recent measurements (October 1997).

During 1997, Woodward-Clyde completed slug tests in monitor wells MW-EDC-4, MW-EDC-13, and MW-EDC-18 to calculate hydraulic conductivity. The hydraulic conductivity values ranged from 0.0004 to 0.000826 cm/sec. The arithmetic average hydraulic_specific conductance for these three wells was 0.000661 cm/sec (1.87 ft/day).

2.4 GROUNDWATER SAMPLING AND ANALYSIS RESULTS

Groundwater was sampled from the temporary well points during the Phase I groundwater investigation and analyzed for nitrate, sulfate, lead and chromium. The Phase II monitor wells and four monitor wells installed previously at the EDCC landfill area were sampled during the Phase II groundwater investigation and analyzed for nitrate, sulfate, pH, specific conductance, temperature, turbidity, chromium and lead. Eight monitor wells were also sampled during October 1997. All sampling results have been previously reported to the ADPC&E.

Groundwater from 10 of the monitor wells have exhibited nitrate concentrations in excess of 10 mg/L. These monitor wells are MW-EDC-6, 7, 8, 9, 10, 11, 14, 15, 16, and 17. As specified in the CAO, these 10 monitor wells are the wells which will be included in the insitu bioremediation program. Soil boring logs and monitor well construction diagrams are presented in Appendix A. Table 2.2 lists the analytical results for nitrate, sulfate, pH, and specific conductance for these 10 monitor wells.

Risk-based target monitoring levels for nitrate in groundwater at the site were developed and submitted to the ADPC&E in a report prepared by Woodward-Clyde in February, 1997. Groundwater modeling and risk assessment procedures were utilized in this work. The findings indicate that the existing nitrate concentrations in shallow groundwater on the site do not present unacceptable risks to human health.

2.5 REQUIREMENTS OF THE CAO

As discussed in Section 1.0, the CAO requires an Interim Measure which includes in-situ bioremediation in the existing groundwater monitoring wells which have exhibited nitrate concentrations in excess of 10 mg/L. Specifically the CAO requires:

• Submission of a work plan describing activities and including a schedule of significant dates for initiation of bioremediation, sampling the groundwater and submission of the sample analysis results to the ADPC&E, and preparation of annual reports evaluating the effectiveness of the Interim Measure.

- Upon receiving the <u>ADPC&E's written approval of the work plan</u>, implementation of the in-situ bioremediation.
- Continuation of the in-situ bioremediation activities until nitrate concentration become less than 10 mg/L or for 12 months after completion of the wastewater improvements required by Paragraph 2 of the CAO, whichever comes first.
- At the completion of the Interim Measure, prepare and submit to the ADPC&E an Interim Measures Report. If nitrate concentration exceeds 10 mg/L for any well at that time, the report is to include a recommendation of a Final Remedy and supporting documentation including an evaluation of the potential risks to human health and the environment associated with the residual concentration of nitrates in the groundwater. A determination may be made that further remediation is not necessary or that additional action may be necessary.

Nitrate can be used as an electron acceptor in microbiologically mediated oxidation reduction reactions. In these denitrification reactions, nitrate is reduced to NO_2^- , N_2O , NO, NH_4^+ , or N_2 and organic matter (which is utilized as the electron donor) is oxidized. Sufficient organic matter to allow significant biodegradation of the nitrate is not expected to be present in the groundwater at the site. Consequently, the approach for the in-situ bioremediation involves adding organic matter (organic supplement) to the monitor wells to enhance the biodegradation of the nitrate.

The in-situ bioremediation is planned to be implemented in three steps as follows:

- Initial sampling and testing. Data will be collected from the ten monitor wells to provide baseline data necessary to design the in-situ bioremediation. The 10 monitor wells will be sampled and the groundwater samples will be analyzed for parameters that identify geochemical conditions, parameters that are degradation products of microbiological mediated oxidation reduction reactions, microbiological parameters, and parameters that are alternate electron acceptors. In addition, slug tests to determine hydraulic conductivity will be completed in each monitor well. These tests will allow design of the quantity and type of organic supplement to be added to enhance in-situ bioremediation and will also provide baseline information which can be used to assess effectiveness of the bioremediation.
 - Preliminary field testing. Based on the results from the initial sampling and testing, two of the monitor wells will be selected to represent the range of conditions present in the ten wells. These selected monitor wells will then be used in <u>short term field tests</u> of the addition of organic supplement. These field tests will be used to verify the amount and mechanism for addition of the organic supplement. Following completion of the field tests, the procedures will be finalized for implementation of the in-situ bioremediation in all 10 monitor wells.
- Implementation of in-situ bioremediation. The enhanced in-situ bioremediation will be implemented in all 10 monitor wells. This will include periodic addition to each well of the organic supplement, periodic sampling and analysis of groundwater from each well, and reporting to the ADPC&E

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results of the analysis. As specified by the CAO, this will continue until nitrate concentrations are less than 10 mg/L in the monitor wells or for 12 months after completion of the wastewater improvements required by Paragraph 2 of the CAO, whichever occurs first. EDCC will then prepare and submit to the ADPC&E an Interim Measures Report, as specified by the CAO.

The above activities are discussed in more detail in Sections 3.1 through 3.3. Section 4.0 discusses the overall schedule for these activities. Section 5.0 discusses reporting throughout the implementation of this plan.

3.1 INITIAL SAMPLING AND TESTING

The 10 monitor wells will be sampled and analyzed one time for the following chemical and microbiological parameters:

- *pH.* Biological activity can be inhibited by extremely low or high pH values. Current pH values will be considered in selecting the organic supplement for enhancing bioremediation. Denitrification is a hydrogen consuming reaction. The groundwater pH is likely to change (become more basic) as denitrification proceeds. Denitrification rates may be low at high pH.
- **Specific Conductance.** Specific conductance provides a general indication of the amount of dissolved solids present and changes in specific conductance during bioremediation may be used in assessing progress.
- **Oxidation-Reduction Potential (ORP).** ORP is a general indicator of the oxidizing or reducing conditions present in the groundwater and can be useful in evaluating reactions that compete with denitrification and in designing the organic supplement.
- **Temperature.** Groundwater temperature affects microbiological activity. Biodegradation rates are generally accelerated at temperatures above 20 degrees C.
- **Chloride.** At high concentrations chloride can inhibit microbiological activity.

- **Alkalinity.** Carbon dioxide released by microbiological activity reacts with carbonate minerals to form bicarbonate, which increases alkalinity. Increases in alkalinity can indicate increased biodegradation.
- **Methane.** Methane is a degradation product of anaerobic biodegradation of many organics and can be a measure of progress of bioremediation.
- **Dissolved Oxygen (DO).** DO is an electron acceptor which competes with nitrate in oxidation reduction reactions. Nitrate will generally not be degraded until DO has been depleted.
- **Carbon Dioxide.** Carbon dioxide can be a degradation product of many organics and is a potential indicator of degradation activity.
- *Nitrate.* The amount of nitrate present is a key factor in designing the addition of organic supplement to enhance in-situ bioremediation.
- *Nitrite.* Nitrite is a potential degradation intermediate of nitrate biodegradation.
- Ammonia and Total Kjeldahl Nitrogen (TKN). Ammonia and organic nitrogen, as measured by TKN, can affect microbiological activity. Ammonia and organic nitrogen could also possibly be intermediate degradation products during bioremediation of nitrate.
- **Ferrous Iron.** Ferrous iron is a product of utilization of ferric iron from the soil as an electron acceptor in microbiological mediated oxidation reduction reactions. However, when present in sufficient concentration, nitrate will generally out compete ferric iron as an electron acceptor in microbiological mediated oxidation reduction reactions. Ferrous iron may be produced during autotrophic denitrification which can often occur with heterotrophic denitrification. Analysis of ferrous iron may be useful as a process control for denitrification.

8:5.1 Autotrophic bacteria

Autotrophic bacteria obtain their nutritive carbon from carbon dioxide, but it is the processes by which they obtain energy that make them beneficial to humans. Specific groups of autotrophic bacteria can oxidize ammonia, nitrites, sulfur, iron, manganese, hydrogen, and carbon monoxide (which also can be reduced by a different bacterium). The oxidation transforms mineral forms, which are often less useful to plants (nitrites, sulfides) to useful forms (nitrates, sulfates). Other oxidations eliminate toxic forms of carbon and manganese.

Probably the most important groups of autotrophic soil bacteria are those that oxidize ammonia to nitrites (a toxic form of nitrogen that is short lived) and then to nitrates (see Sec. 6: 3.3). These *nitrifying* organisms achieve maximum growth under the following conditions:

1 The presence of proteins to release ammonia as they decompose, or the presence of ammonia salts, such as ammonium sulfate.

- 2 Adequate aeration.
- 3 A moist but not wet soil.
- 4 A large amount of calcium (not strongly acidic).
- 5 Optimum temperature at 99°F (37°C) with a minimum of 41°F (5°C) and a maximum of 131°F (55°C).

The bacterial nitrification process is shown diagrammatically as follows:

	Bacteria		Bacteria	
NH4 ⁺	}	NO_2^{-}	>	NO ₃ -
ammonium		nitrite		nitrate
(utilized by		(temporary)		(utilized
plants)		(apparently		by plants)
		(OAIG)		

Nitrification is of great concern to the quality of our environment because the conversion of stable NH_4^+ into NO_3^- may mean that the nitrates will move to contaminate groundwaters. Although a high nitrogen (protein) content is usually desirable, nitrates in high concentrations are toxic to humans and animals; in lesser concentrations, nitrates in streams and lakes can promote the growth of algae (eutrophication), which, when dead and decomposing, consume water oxygen needed by aquatic faunae.

Several groups of autotrophic bacteria can either oxidize or reduce carbon monoxide to carbon dioxide or methane, all of which are gases.⁷ These bacteria are literally life giving. The world's population adds 220 million tons (200 million mt) of carbon monoxide to the atmosphere each year. With no conversion to carbon dioxide or methane, the atmosphere would become lethal to animal life within a few years. The bacterial conversion is always anaerobic (the absence of free oxygen), but in the absence of hydrogen, carbon monoxide (CO) is oxidized to carbon dioxide (CO₂); in the presence of hydrogen, CO is reduced to methane (CH₄). The CO oxidizing and reducing bacteria are present

⁷Robert E. Inman, Royal B. Ingersoll, and Elaine A. Levy, "Soil: A Natural Sink for Carbon Monoxide," *Science*, 172 (18 June 1971), 1229-31.

in most soils but not in those under asphalted surfaces or in congested central cities. This is one more argument for establishing and maintaining green belts to enhance the environment near large concentrations of people.

Denitrification, the reduction of nitrates (NO_3^-) to inert nitrogen gas (N_2) , is another autotrophic bacterial reaction that is anaerobic (see Sec. 6: 3.5). *Thiobacillus denitrificans* is one of the more common bacteria of this type. The presence of elemental sulfur hastens denitrification.

8:5.2 Heterotrophic bacteria

Heterotrophic bacteria are those that depend upon organic matter for their nutrition; in this group are most of the soil bacteria. Heterotrophic bacteria include both nitrogen-fixing and non-nitrogen-fixing kinds; the nitrogen-fixing bacteria groups are subdivided into symbiotic and nonsymbiotic. The symbiotic bacteria are most commonly associated with leguminous plants, but some nonleguminous plants also have symbiotic nitrogen-fixing bacteria.

Heterotrophic bacteria, which do not fix nitrogen, are the most prevalent type of these common soil bacteria. The non-nitrogen-fixing bacteria are the bacteria most responsible for the decomposition of organic matter. Some members of this group include (1) the *thermophilic* bacteria that continue decomposition even when released energy heats and occasionally burns hay when it is baled too wet, and (2) certain of the genus *Bacillus*, some of which produce antibiotics.

8:5.3 Symbiotic bacteria

The word **symbiotic** comes from the Greek and means "living together." It is used to describe the heterotrophic bacteria that fix atmospheric nitrogen in plant root nodules because it illustrates the dependence of the process on *both* the bacteria and the host plants, the majority of which are legumes.

In the days of ancient Greece, it was known that legumes (pod-bearing plants such as peas, beans, alfalfa, and clovers) had a beneficial effect upon whatever crop was next planted in the same soil. It was not until 1838 that Boussingault demonstrated that the beneficial effect was due to the fixation of atmospheric nitrogen in the legume root nodules. In 1879, Frank proved that artificial inoculation with specific bacteria resulted in legume root nodule formation and that the bacteria in these nodules fixed atmospheric nitrogen.

Symbiotic bacteria attack the root hairs of the host plant, and the resulting injury induces the root cells to grow around the bacteria, thus forming root bulges called nodules. The bacteria live within these nodules that are produced. The plant roots supply essential minerals and organic matter to the bacteria and eventually benefit from the atmospheric nitrogen that the bacteria have fixed and used to build body protein.

- **Total Iron.** Total iron may be a source of autocatalytic denitrification. Suspended solids present in the well water may provide a source of nonferrous iron.
- **Manganese.** Manganese is a potential electron acceptor which may compete with nitrate in microbiological mediated oxidation reduction reactions.
- **Sulfate.** Sulfate is a potential electron acceptor. However, nitrate, when present in sufficient concentration, will generally out compete sulfate as an electron acceptor in microbiological mediated oxidation reduction reactions.
- **Total Organic Carbon (TOC).** TOC is a general indicator of the amount of organic matter present and is useful in determining the amount of organic supplement that will be required to enhance the bioremediation. Determination of current TOC concentrations can allow monitoring of TOC during bioremediation as a surrogate analyte for the organic supplement.
- **Total Phosphate.** Phosphate is an essential nutrient and, if not naturally present in sufficient concentrations, will need to be present in the organic supplement.
- **Microbiological Testing.** For denitrification, denitrifying bacteria can be determined. Denitrifying bacteria are a heterogeneous group of organisms that mediate the conversion of nitrate to elemental nitrogen. Testing for denitrifying bacteria can be useful in determining if the appropriate microbial population of microorganisms are being enhanced and if the injection of nutrients is stimulating microbial activity.

3.1.1 Sampling Procedures

The monitor wells will be purged and sampled using a redi-flow submersible pump. The general procedure will be as follows:

The well caps will be removed and the static water level and the total depth of the well will be measured using an electronic water level sensor. Depth to water will be referenced to the survey point (generally the top of the PVC well casing).

Wells will be purged until a volume equaling at least three times the calculated standing water in the well has been removed. This is calculated as follows:

Volume of water in the casing (gallons) = $3.14 \text{ x } \text{r}^2 \text{ x}$ (H-h) x 7.48

where:

Depth to water (feet) = h Depth of well (feet) = H Radius of well (feet) = r

The specific conductance, pH, temperature, DO, and turbidity will be measured prior to and during purging. If the specific conductance (within 10%), pH (within 0.1 standard units), temperature (within 0.5 degree C), DO (within 10%) or turbidity within (10% or less than 5 NTU) of the water has not stabilized when three well volumes have been purged, then purging will continue until these parameters have stabilized or to a maximum of five well volumes. If a well is purged to dryness, the well will be allowed to recover prior to collecting samples.

When the well has been purged the samples will be collected with the redi-flow pump. The sampling pump will be decontaminated between each use. New discharge tubing will be used at each well. Discharge rate of the pump will be reduced to approximately 100 ml/min for sampling for dissolved gases. Samples for analysis for ferrous iron will be filtered with a 0.45 micron filter placed in-line on the discharge line from the pump. All other samples will be unfiltered. Flow-through cells will be used for field measurements of samples for pH, specific conductance, DO, ORP, and temperature.

3.1.2 Analysis of Samples

The analytical methods, containers, and preservatives for each analyte are listed in Table 3.3.1. Samples will be handled under chain-of-custody procedures and will be transported to a commercial laboratory for analysis.

3.1.3 Hydraulic Conductivity Tests

Slug tests to determine hydraulic conductivity will be completed for each of the ten monitor wells. The following procedures will be used:

- The static water level will be measured using an electronic water level sensor which will be decontaminated between wells.
- A pressure transducer will be lowered into the well to measure water level changes and proper functioning of the transducer and recording data logger will be checked.
- A slug (suspended weight of known volume) will be lowered into (slug-in test) the water in the well. The slug will be introduced as rapidly as possible without causing undue splashing or turbulence, so as to produce an instantaneous head increase. The magnitude of the slug will be such that the slug will not cause water to overflow the well casing. The slug will be decontaminated prior to use in each well.
- The data logger will record the water levels continuously (millisecond intervals). Measurements will continue until the water level stabilizes.
- Water level measurements will be checked for credibility before proceeding to the next testing stage (slug-out test).
- When checking of the data has been completed and the data determined to be acceptable, the slug will be removed from the well as rapidly as possible, so as to produce an instantaneous head reduction. Water levels will be measured and recorded continuously, the same as in the slug-in stage of the testing. Measurements will be terminated when the water level stabilizes.
- Water levels will be checked for credibility before proceeding to the next well. Any test or test stage will be repeated if the field review of the recorded data indicates questionable validity.

• The data from the slug test will be reduced to evaluate time versus change in water level using the methods of Cooper et. al. (1967) or Bouwer and Rice (1976).

3.2 PRELIMINARY FIELD TESTING

The data collected during the initial sampling and testing will be used to design the preliminary field tests. This will involve selection of two or three monitor wells for the tests, selection of the appropriate organic supplement, and selection of procedures for supplement addition and for monitoring. These activities are discussed in Sections 3.2.1, 3.2.2, and 3.2.3, respectively.

3.2.1 Selection of Monitor Wells

Monitor wells will be selected to provide tests representative of the range of conditions present in the 10 monitor wells. Key factors in the selection are expected to be concentrations of the various nitrogen species (nitrate, ammonia, and TKN), pH, and microbiological parameters.

Final determination of the wells will be made after the initial sampling and testing has been completed. It is anticipated that two pilot wells will be used to evaluate denitrification potential. Two ranges of nitrate concentrations will be evaluated: 10 - 400 mg/L and above 400 mg/L. Denitrification in the low concentration range is less likely to be complicated by clogging of the permeable zone due to biomass growth and the production of intermediates such as nitrite. However, areas of high nitrate concentration are more likely to develop denitrifying microbial populations. The optimum pH range for denitrification is 7 - 8 (Bengtsson 1995); however, since denitrification is a hydrogen consuming reaction, denitrification may be preferred at low pHs.

3.2.2 Selection of Organic Supplement and Application Method

There are numerous methods that are utilized in the denitrification of groundwater. For insitu denitrification, two common practices are:

1. Direct injection of a carbon source into selected wells. The most direct and straightforward method of promoting denitrification is to inject wells with a

carbon substrate. Carbon limitation is often the cause of nitrate accumulation in groundwater (Bradley 1992). A variety of common substrates may be utilized, such as ethanol, methanol, sucrose, molasses, cellulose, acetate, and The carbon sources that are most readily utilized by indigenous urea. microorganisms in the groundwater are natural carbon sources, such as cellulose and acetate; however, positive results have been achieved with a variety of carbon sources. Molasses, methanol, cellulose, and acetate will be evaluated following completion of the initial sampling and testing. One or two of these supplements will be selected for use in the field tests.

The most significant drawback to substrate injection is aquifer fouling due to the production of biomass (Hiscock et.al.1991 and Dahab 1993). In some cases, significant denitrification can lead to gas production (nitrous oxide or nitrogen) that can also promote aquifer clogging. To limit the potential for aquifer clogging, the substrate may be pulse injected. Initially, the concentration of substrate will be limited to reduce the potential for over production of biomass.

2. Utilization of Hydrogen Releasing Compound[®]. Hydrogen Releasing Compound (HRC) is produced by Regenesis to promote reductive dechlorination of chlorinated solvents. The HRC, upon hydration, slowly releases lactic acid into the groundwater. The lactic acid can be fermented to produce hydrogen gas which drives reductive dechlorination. In the presence of other electron acceptors, such as nitrate and sulfate, HRC is utilized by denitrifying and sulfate reducing bacteria. The principal advantage of HRC is $yr^{a^{i}}$ reduced potential for aquifer clogging versus direct injection. HRC may be evaluated for wells where aquifer clogging due to biomass production is observed.

Direct injection of molasses, methanol, cellulose, or acetate is anticipated to be the preferred remedial method. HRC may be evaluated if aquifer clogging is observed. Additional nutrient needs (i.e. ammonia and/or phosphate) will be evaluated during the preliminary field testing.

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3.2.3 Procedures for the Tests

Procedures for the tests will be finalized after completion of the initial sampling and testing. However, these tests are expected to be conducted for an approximately eight week period. The organic supplement will be periodically added to each of the selected wells during the test period. The frequency and amount of addition is dependent on findings from the initial sampling and testing. Prior to each addition of supplement and at the end of the test, each well will be sampled and analyzed for certain parameters. On a preliminary basis these parameters are expected to be the following:

					Wee	k			
Parameter	0	1	2	3	4	5	6	7	8
pН	x	X	X	x	X	X	X	X	X
Specific Conductance	X				X				X
DO	X	X	X		X				X
ORP	x		X	1	X				X
Nitrate	x	X	X	X	X	x	X	X	X
Ammonia	X		X		X		X		x
TKN	x				X				X
TOC	X		X		X		x		x
Alkalinity	x				X	-			X
Temperature	X	X	X	X	X	x	X	x	x
Carbon Dioxide	X		x		X	-	x		X
Nitrite	X		X		X		X		x
Chloride	X				X				x
Phosphate	X	1	x		X		X		x
Manganese	x				X				x
Denitrifying Bacteria	X		X		X				x
Ferrous Iron	X				X				X
Total Iron	x		1	1	X				x

Many of these analytes can be analyzed in the field to provide immediate data for evaluating supplement addition rates.

3.3 IMPLEMENTATION OF IN-SITU BIOREMEDIATION

The specifics of the bioremediation will be developed following completion of the preliminary field tests. This information will be transmitted to the ADPC&E prior to implementing the in-situ bioremediation.

Specifications for the mix, frequency, and method of addition of organic supplement will be provided for each of the 10 monitor wells. Each monitor well will be inspected each time supplement is added.

A monitoring program will be specified as part of the implementation. Specifics will be presented in the aforementioned transmittal to the ADPC&E. Semiannual monitoring of the 10 monitor wells for a specific set of analytes is anticipated. Analysis for additional parameters may be specified on an annual basis, as necessary. Analytical results from each sampling will be reported to the ADPC&E within 45 days of receipt by EDCC. Results will be discussed in the annual report to the ADPC&E and in the Interim Measures Report.

Figure 4-1 presents the proposed overall schedule. Milestones include the following:

- December 10, 1998 Submittal of work plan to ADPC&E
- January 29, 1999 Anticipated receipt of ADPC&E approval of work plan
- ≈ March 29, 1999 Submittal of results of initial sampling and testing and of plans for preliminary field tests
- ≈ August 29, 1999 Submittal of results of preliminary field tests and of plans for implementation
- ≈ September 29, 1999 up to August 1, 2002 Operation of in-situ bioremediation
- November 1, 2002 Submittal of Interim Measures Report

The above milestones are based on an assumed date of the ADPC&E approval of the Work Plan and the schedule may vary according to the actual date of approval. Also, the assumed period of operation of the in-situ bioremediation and of submittal of the Interim Measures Report may be accelerated if the nitrate concentrations are reduced before the latest date for completion of operation (August 1, 2002) is reached. The following reports are planned:

- The results of the initial sampling and testing and the plans for the preliminary field tests will be transmitted as a letter report. All analytical results will be appended. Relevant findings will be discussed.
- The results of the preliminary field tests and the plans for the implementation of the in-situ bioremediation will be transmitted to the ADPC&E as a letter report. All analytical results will be appended. The planned supplement mix and the frequency and amount of addition for each well will be identified. The monitoring and reporting program will be specified.
- During implementation of the in-situ bioremediation, monitoring results will be submitted to the ADPC&E within 45 days of receipt of laboratory results. Monitoring is expected to be on a semiannual basis.
- Annual reports for a calendar year will be submitted by March 31 of the subsequent year. The reports will include all analytical results for the reporting period. Concentration changes over time will be discussed. Changes in the remediation will be discussed, if any changes were necessary. A groundwater elevation map will be included and any significant changes in direction of hydraulic gradient will be discussed.
- An Interim Measures Report will be submitted to the ADPC&E following completion of the Interim Measures. The report will document the Interim Measures. If nitrate concentration in any monitor well exceeds 10 mg/L, an evaluation of risks to human health and the environment associated with the residual concentrations of nitrate will be included. The risk evaluation will utilize the factors identified in the ADPC&E Regulation 22 at Section 22.1207(c)(1) through 22.1207(d)(8). The report will either include a recommendation that further action is not necessary or it will include a recommendation and supporting documentation for a Final Remedy.

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- Woodward-Clyde, 1997. Phase II Groundwater Assessment Report, El Dorado Chemical Company, El Dorado, Arkansas. January 1997.
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TABLES



TABLE 2-1

DESCRIPTION OF REGIONAL HYDROGEOLOGIC UNITS IN THE EL DORADO, ARKANSAS AREA EL DORADO CHEMICAL COMPANY EL DORADO, ARKANSAS

System	Series	Group	Formation	Hydrogeologic Unit	Hydrogeologic Properties
Quaternary	Holocene and Pleistocene		Alluvial and terrace deposits		Clay, silt, sand, and gravel. Present only in bottomlands of most streams. Generally not used. As much as 100 feet thick.
			Cockfield Formation	Cockfield aquifer	Lignitic sand with interbedded clay. Principal aquifer for rural domestic supply. Approximately 200 feet thick where present.
			Cook Mountain Formation	Cook Mountain confining unit	Clay with interbedded fine sand. Not an aquifer. Thickness ranges from 50 to 200 feet.
Tertiary	ary Eocene Claiborr	Claiborne		Greensand aquifer	Thinly bedded fine glauconitic sand with interbedded clay. Source of municipal and industrial water supply principally in southeast part of county. Water withdrawals approximately 0.5 million gallons per day. Approximately 200 feet thick.
			Sparta Sand	Middle confining bed	Clay and silt. Not an aquifer. Thickness ranges from 40 to 160 feet.
				El Dorado aquifer	Thickly bedded medium to coarse sand. Source of municipal and industrial water supply throughout the county. Water withdrawals approximately 14 million gallons per day. Approximately 300 feet thick.
			Cane River Formation	Cane River confining unit	Clay and silty clay. Not an aquifer. Approximately 300 feet thick.

From Leidy and Taylor, 1992

TABLE 2-2

ANALYTICAL RESULTS MONITOR WELLS THAT HAVE EXHIBITED NITRATE CONCENTRATIONS EXCEEDING 10 MG/L EL DORADO CHEMICAL COMPANY EL DORADO, ARKANSAS

Well No.	Nitrate (mg/l) March-April 1996	Nitrate (mg/l) October 1997	Sulfate (mg/l) March-April 1996	Sulfate (mg/l) October 1997	pH (S.U.) March-April 1996	рН (S.U.) October 1997	Specific Conductance (umhos/cm) March-April 1996	Specific Conductance (umhos/cm) October 1997
MW-EDC-6	51.1	NS	24	NS	7.7	NS	700	NS
MW-EDC-7	282	NS	380	NS	8.1	NS	700	NS
MW-EDC-8	1010	773	68.3	95.9	7.9	4.3	710	735
MW-EDC-9	37.3	NS	621	NS	9	NS	320	NS
MW-EDC-10	257	NS	89	NS	7.7	NS	410	NS
MW-EDC-11	22.1	NS	578	NS	11.1	NS	130	NS
MW-EDC-14	11.9	NS	139	NS	4.6	NS	650	NS
MW-EDC-15	34.5	NS	4.4	NS	6.4	NS	266	NS
MW-EDC-16	137	NS	4.6	NS	5.7	NS	890	NS
MW-EDC-17	45	109	145	74.5	4.9	4.2	730	1230

NS = NOT SAMPLED

TABLE 3-1

SAMPLE CONTAINERS, PRESERVATION METHODS AND ANALYTICAL METHODS EL DORADO CHEMICAL COMPANY EL DORADO, ARKANSAS

		Field or					
	Analytical	Laboratory		,	Field		Special
Parameter	Method/Instrument	Analysis	Container	Preservation '	Filtered	Hold Time *	Consideration
	Multi-Parameter Probe						Two-point calibration between wells;
Dissolved Oxygen	with Flow-Through Cell	Field	NA	None	No	NA	record parameters when readings stable.
	Multi-Parameter Probe						
Oxidation-Reduction Potential (ORP)	with Flow-Through Cell	Field	NA	None	No	NA	Record parameters when readings stable.
	Multi-Parameter Probe						
Specific Conductance	with Flow-Through Cell	Field	NA	None	No	NA	Record parameters when readings stable.
	Multi-Parameter Probe						
Temperature	with Flow-Through Cell	Field	NA	None	No	NA	Record parameters when readings stable.
	Multi-Parameter Probe						
рН	with Flow-Through Cell	Field	NA	None	No	NA	Record parameters when readings stable.
Ferrous Iron	HACH Test Kit-8146	Field	1 x 100 mL (G,P)	Ice (4°C)	Yes	ASAP	Minimize introduction of oxygen.
Carbon dioxide	HACH Test Kit 26193-07	Field	1 x 100 mL (G)	None	No	ASAP	
Total Iron	EPA Method 3010, 6010	Laboratory	1 x 250 mL (G,P)	Cool 4°C, HCl to pH<2	No	6 Months	
Manganese	EPA Method 3010, 6010	Laboratory	l x 250 mL (G,P)	Cool 4°C, HCl to pH<2	No	6 Months	
Major Anions (Cl [*] , NO ₃ [*] , NO ₂ [*] , SO ₄ ^{*2*})	EPA Method 300, 353.2	Laboratory	1 x 1,000 mL (G,P)	Ice (4°C)	No	48 Hours	
Alkalinity	EPA Method 310.1	Laboratory	1 x 250 mL (G,P)	Ice (4°C)	No	14 Days	
	Method AM-19						Method by Microseeps, Inc.
Methane	or RSK-175	Laboratory	3 x 40 mL VOA	lce (4°C)	No	ASAP	or Robert S. Kerr Laboratory
Ammonia	EPA Method 350.3	Laboratory	1 x 1,000 mL (G,P)	Cool 4°C, H2SO4 to pH<2	No	28 Days	
Total Kjeldahl Nitrogen	EPA Method 351.3	Laboratory	1 x 1,000 mL (P)	Cool 4°C, H2SO4 to pH<2	No	28 Days	
Total Phosphate	SW846 9060	Laboratory	1 x 500 mL (P)	Cool 4°C, H2SO4 to pH<2	No	48 Hours	
Total Organic Carbon	SW846 9060	Laboratory	2 x 125 mL (A,G)	Cool 4°C, H2SO4 to pH<3	No	28 Days	
Denitrifying Bacteria	HACH Test Kit 1436-01	Field	50 mgL (G)	None	No	ASAP	

NOTE:

Wells to be purged and sampled as per USEPA Region IV and DNR Protocols.

Key:

NA = Not Applicable. G,P = Glass or Polyethylene container.

A = Amber container.

SW846 = Test Methods for Evaluating Solid Waste, dated 4/92.

EPA = "Methods for Chemical Analysis of Water and Waste", 1979, Revision 3/83.

¹ Sample preservation is performed by sampler immediately upon sample collection if not already done so by the laboratory.

² Hold time based upon day of sample collection not verified time of sample receipt.

FIGURES





2-1



FIGURE 4-1. PROPOSED SCHEDULE IN-SITU BIOREMEDIATION FROM EL DORADO CHEMICAL COMPANY EL DORADO, ARKANSAS



TACK DECODIDION		1								1							C	alenc	ar	IIme	(мопт	ns)					
TASK DESCRIPTION	DECEMBER	1999 January feruary	MARCH API	Ril WAS	JUNE	ULY AUGU	ST STPTEWRA	OCTOBIA NUMEN	ABER : DECEMBER	2000 I January i feb	aunay ² m	ARCH APR		Y	F JB	iy MiG	WST SEP	ENGER OCTOR	FF NOVEMB	FR - OFCEMBER	2001 January feer	WARY	NARCH APRIL	MAY	: .8 H	YIIY	AUGUST SF
Submission of Work Plan			1	I								1					1										1
ADPCE Approval of Work Plan		4	1		4 .												1	i.							1		I
Initial Sampling and Testing				1								COME FOR M		i	_			:		1	1		¢ ¢				1
Submission of Letter Report							1			2					·			1	Constanting of the			I		<u>.</u>	1		1
Preliminary Field Testing							-			1	<u>-</u>		1	:				I				Ì	,		1		-
Submission of Letter Report				:		4	Å		1		i				i		2			1							1
Implementation of In-Situ Bioremediation						l		· · · · · · · · · · · · · · · · · · ·									į					{				1	,
Semiannual Analytical Results Submissions							1			: 1					ļ			İ				1	1			· · ·	
Annual Reports			1	1			1							1	:			· · ·	^j		·		4		:	ε Ι	
Interim Measure Report			-	1		1	!								Ť						, F		1	1			1

URS Greiner Woodward Clyde



APPENDIX A

SOIL BORING LOGS AND MONITOR WELL CONSTRUCTION DIAGRAMS FOR MONITOR WELLS TO BE USED IN THE IN-SITU BIOREMEDIATION

Nov 11 98 11:55a

Monitor W	/ell Installat	ion	BORING	MARTICAG
24 DA1800	, Arkansas		DATE:	95B165 2/21/96
El Dorado El Dorado Anderson	Chemical C , Arkansas Engineering	ompany	GEOLOGIST: APPROVED: PAGE:	EJF LOC 1 of 1
Hollow-ster	n Augered:			
SAMPLE				
S.P.T.(b/ft) or P.Pen.(uf)	Recovery (inch)	Description	of Stratum	
	18	Light gray (SYR, 7/1) Clayey SILT; damp; d	ense.	(ML)
	18	Reddish gray (5YR, 5/2) Silty CLAY; dry to	damp.	(CL)
	18	Wet reddish gray (5YR, 5/2) SILT and Silty S	SAND.	(ML/SM)
		End of Boring at 19'.		
		- ·		
		6 Woodward-Clude Commu	tants	
	El Dorado El Durado Anderson Hollow-ster	El Dorado Chemical C El Dorado, Arkansas Anderson Engineering Hollow-stem Augered: S.P.T. (b/ft) or P.Pen.(uf) I8 I8 I8 I8 I8 I8	El Dorado Chemical Company El Dorado, Arkanses Anderson Englineering S.P.T.(M) Recovery Description P.P.m.(ut) (inch) Description 18 Light gray (SYR, 7/1) Clayey SILT; damp; d 18 Reddish gray (SYR, 5/2) Silty CLAY; dry to 18 Wet reddish gray (SYR, 5/2) SILT and Silty 3 18 Wet reddish gray (SYR, 5/2) SILT and Silty 3 End of Boring at 19'.	EI Dorado, Arkanas Anderson Engineering Billow-sem Augenet: S.F.T. (Ant) S.F.T. (Ant) Recovery P.Pen.(ut) IS Ight gray (SYR, 7/1) Clayey SILT; damp; dense. IS Ight gray (SYR, 5/2) Silty CLAY; dry to damp. IS Keddish gray (SYR, 5/2) Silty CLAY; dry to damp. IS Keddish gray (SYR, 5/2) Silt and Silty SAND. End of Boring at 19°.

p.7



v 11	98 11:5	56a		1-501-223-2996	5			
ROJECT: DCATION: LIENT: RILLER:	Monitor V El Dorado El Dorado El Dorado Anderson	Monitor Well Installation BORING: El Dorado, Arkansas PILE: El Dorado Chemical Company GEOLOGIST: El Dorado, Arkansas APPROVED: Anderson Engineering PAGE:						
(FEET) SYMBOL	Hollow-siz	m Augered:						
0	·\$.P.T.(b/ft) or P.Pcn.(tsf)	Recovery (inch)	Desc	cription of Stratum				
1		18	Reddish yellow (7.5YR, 6/6) Silty SA	ND with traces of light gray SILT; dry	4- (SM			
		18	Stiff reddish gray (5YR, 5/2) CLAY.		(CI			
		18	very stiff dark gray (10YR, 4/1) CL	AY.	(CI			
8-1		18	Wet, loose light gray (5YR, 7/1) SILT	and fine Silty SAND.	(ML/SM			
2			End of Boring at 22'.					
-	kal	I I.						
N 4 96 ENV.	18728 58165 MW		Woodward-Clyde Co	nsultants				

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r				LOG OF BORING		
PROJEC	.T: 10N:	Monitor W El Dorado	/ell Installat , Arkansas	ion	BORING: HLL:	MWEDC-08 95B165
CLIENT	ſ:	El Dorado	Chemical C	ompany	DATE: GEOLOGIST	2/20/96 EJF ()
DRILL	ER:	Anderson	, Arkansas Engineering		PAGE:	1 of 1
		Hollow-ster	n Augered:			
DEPTH (FEET)	SYMBOL					
		S.P.T.(b/ft) or P.Pen.(usl)	Recovery (inch)	ľ	Description of Stratum	
- 0 - 1- - 2- - 3- - 4- - 5- - 6- - 7-			18	Fill material consisting of dark bro CLAY, yellow brown medium SA	wn (5YR, 7/1 + 10YR, 6/3) ND, and light gray Silty SAN	Sandy (SC/SP) D, loose.
- 8 - - 9 - - 10 - - 11 -			18	Light gray (5YR, 7/1) Silty CLAY	; medium dense; damp.	(CH)
- 12 - - 13 - - 14 - - 15 - - 16 - - 17 - - 18 -		Les	18	Very stiff dark grayish brown (10)	(R, 4/1) CLAY.	
- 19 - - 20- - 21 - - 22 -			. 4	Light gray (SYR, 7/1) CLAY and	light brown fine SAND.	(CL/SM)
- 23 - - 24 - - 25- - 26 -				—wet, light gray (5YR, 7/1) fine S	Silty SAND.	(SM)
- 27	191			End of Boring at 27'.		
	-	····	-			
				G Woodword Chile	Conquitanta	



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PROJECT: LOCATION: CLIENT:	Monitor W El Dorado, El Dorado	ell Installat Arkansas Chemical C	on pady	boring: M File: 955 Date: 2/1 Geologist: Ej	WEDC-09 8165 5/96 F()
DRILLER:	El Dorado, Anderson l	Arkansas Engineering		Approved: JL PAGE: 1 c	£ f1
LEPTH FEET) YMBOL	Hollow-sten	Augered:			
	S.P.T.(b/ft) of P.Pen.(uf)	Recovery (mch)	Descript	ion of Stramm	
1		18	White - light gray (5YR, 8/1-7/1) dense S	ilty CLAY; dry.	(CL)
9-10-11- 11- 12-		18	Light gray (5YR. 7/1) stiff CLAY; damp;	with lepses.	(CL/ML)
13 14 15 16 17	The second se	18	-stiff gray (SYR, 7/1) CLAY; moist		(CL)
18 - 19 - 20 - 21 - 22 - 23 -		18	dark gray (SYR, 7/1) stiff CLAY with	white to light gray SILT; damp	. (CL)
24 - 25 - 26 - 27 - 28 -		18	Wet gray (SYR, 7/1) SILT. End of Boring at 28'.		(MĽ)
			-		
<u> </u>	· .	- <u> </u>			

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Nov 11 98 11:56a

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				LOG OF BORING			
PROJEC	T: ION:	Monitor W El Dorado	Veil Instaliat , Arkansas	ion	BORING: FILE: DATE:	MWEDC-10 95B165 2/19/96	
CLIENT		El Dorado El Dorado	Chemical (, Arkansas	ompany	GEOLOGIST: APPROVED: PAGE	EJF All 1 of 1	
		Hollow-ster	n Apered:				
DEPTH (FEET)	SYMBOL		low-sam Augeren:				
		S.P.T.(b/R) or P.Pen.(tsf)	Recovery (inch)	Description	n of Stratum		
- 0			16	Stiff gray (5YR, 6/1) CLAY; dry.			(CL)
- 9 - - 10- - 11 - - 12 - - 13 -		3	18	Stiff reddish gray (5YR, 5/2) Silty CLAY w dry to damp.	ith traces of yellow Silt	y SAND;	(CL)
- 14 - - 15 - - 16 - - 17 - - 18 - - 19 - - 20 -			18	Reddish brown (5YR, 5/2 + 7/1) CLAY wi and yellow fine sand lenses; wet. End of Boring at 20	th intermitient light bro	wn SILT (CI	_/SM)
	والمحافظ						
				Woodward-Clyde Consu	ltants		



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PROJECT: LOCATION: CLIENT: DRILLER:		Monitor W El Dorado El Dorado El Dorado Anderson	Vell Installat , Arkansas Chemical C , Arkansus Engineering	ompany	BORING: FILE: DATE: GEOLOGIST: APPROVED: PAGE:	MWEDC-11 95B165 2/19/96 EJF MLCL 1 of 1
DEPTH (FEET)	SYMBOL	Hollow-star	n Augered;	an a		
- 0		S.P.T.(b/ft) or P.Pen.(15f)	Recovery (inch)	Descripti	ion of Stratum	
- 1 - 2 - 3 - - 5 - - 5 - - 6 - - 7 -			18	Loose, damp, light gray (SYR, 7/1) and ta	n Silty CLAY.	(CL
9 - - 10- 11 - 12 -			18	Stiff red-gray (SYR, 5/2) CLAY; with inte gray silt lenses; damp to moist.	ermittent tan fine sand an	d light (CL7ML
14 - 15- 16 - 17 -			18	Light gray, tan and reddish brown (10YR, with silty sand lenses.	6/6) fine-medium SAND	; wet; (SM
18 - 19				End of Boring at 19.		
, <u></u>				·		
11 mm				Woodward-Chude Cong	ultante	



Nov 11 98 11:57a

PROJECT: LOCATION: CLIENT: DRILLER:	Monitor W El Dorado El Dorado El Dorado Anderson	/ell Installat , Arkansas Chemical C , Arkansas Engineering	ion Company	BORING: FILE: DATE: GEOLOGIST: APPROVED: PAGE:	MWEDC-14 95B165 2/13/96 E.F. 1 of 1			
DEPTH (FEET) SYMBOL	Hollow-stem Augered:							
0	S.P.T.(b/ft) or P.Pen.(1sf)	Recovery (inch)	Descrip	tion of Stranım				
1	· · · · ·	17	Fine Silty SAND; light gray and tan in co	lor (7.5YR, 7/0).	(SM)			
8		18	Silty CLAY with traces of silt and fine sa (7.5YR, S/0); damp.	nd; light to dark gray in c	olor (CL)			
12	· · · · · · · · · · · · · · · · · · ·	18	Wet fine SAND, primarily gray with trace 6/0-7/0) from sand.	es of tan and light brown	(7.5YR, (ML)			
17	• •• •• •• ••		End of Boxing at 17'.		·			

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JECT:	Monitor V El Dorado	Vell Installat , Arkansas	ion	BORING: FILE: DATE:	MWEDC-15 95B165 2/13/96	
INT: LLER:	El Dorado El Dorado Anderson	Chemical C , Arkansas Engineering	ompany	GEOLOGIST: APPROVED: PAGE:	EIF 1 of 1	
VMBOL	Hollow-ste	m Augened:				
S	S.P.T.(b/ft) or P.Pen.(tsf)	Recovery (inch)	Desc	ription of Stratum		
┍┷╼┺╺┶┺╼┺╼┺╼┺╼ ╾╾╾╾┶╺┶╸╸╸╴╴		18	Fine Silty SAND; damp; gray (7.5YR	, 7/0); traces of red and gray	clay. (S	SM
		18	Light gray to dark gray (7.5YR, 7/0) (silty sand; wet; dense.	Clayey SILT, with traces of 1	ight brown (1	MÍ
		18	Clayey SILT with lenses of coarse s dark brown (7.5YR, 710).	and and gravel; wet; dense; ;	eray with (ML/C	ЗM
			End of Boring at 15".			
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		······	LOG OF BORING	\$0\$31C	MWEDC-16
DIECT: CATION:	Monitor V El Dorado	Vell Installat , Arkansas	ion	BORING: FILE: DATE:	95B165 2/12/96
ENT: LLER:	El Dorado El Dorado Anderson	Chemical C , Arkansas Engineering	ompany	APPROVED PAGE:	1 of 1
ABOL	Hollow-ste	m Augered:			
NAS NAS	S.P.T.(b/ft)				
	or P.Pen.(111)	Recovery (inch)	Descript	ion of Stratum	
∟ - 2 -]					
	- ,	15	Moist fine SAND and Silty SAND; tan -	white in color (7.5YR, 7/7). (SM
		×			
		10	Was Sandy and Silts Of A'd, why is call	P 500 710 712	(0)
		12	wet sandy and sinty CLAY; gray in color	(/.JIR, //0-//0).	(CL
5 - 1 - 1 5 - 1 - 1		20	Wet fine SAND and SILT, stiff; gray in c	olor (7.5YR, 6/0).	(SM
;- 			End of Boring at 17.		

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			LOG OF BORING -		
DIECT: CATION:	Monitor W El Dorado,	'ell Installat , Arkansas	ion	BORING: FILE: DATE:	MWEDC-17 95B165 2/13/96
IENT: ILLER:	El Dorado El Dorado, Anderson l	Chemical C Arkanses Engineering	ompany	GEOLOGIST: APPROVED: PAGE:	LIF 1 of 1
FEET) YMBOL	Hollow-stea	a Augered:			
o is	S.P.T.(b/ft) or P.Pcn.(115)	Recovery (inch)	Descrip	tion of Stratum	
		2	Red (10YR,4/6) Sandy CLAY.		(SC)
0 7 8 9 0 1		18	Pinkish gray (3YR, 6/2) fine Silty SAND	with trace of gray clay.	(SM)
2		12	—light gray (5YR, 6/2) fine Silty SAND;	dry; with a trace of tan S	AND.
9		18	light gray (5YR, 7/1) fine SAND and S fine sand and dark gray clay.	Silty SAND, with traces of	yellow
4		18	—light gray (5YR-7.5YR, 7/0) fine SAN	D; damp.	
9 0	L	18	light gray (7.5YR, 7/0) fine SAND and	l silt; wet.	
4-11	<u>}</u>		End of Boring at 34'.		
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